

# Section 1: Identification of the Substance/Mixture and of Supplier

Product name:	POOL CHLORINE GRANULES
Recommended use: Supplier: Street Address:	Swimming pool water sanitizer Space Industries Limited 160 Plunket Ave, Wiri, Auckland New Zealand
Telephone Number: Facsimile: E-mail: Website: Emergency Telephone Date of preparation:	+ 64 9 262 3902 + 64 9 262 3948 <u>orders@spaceindustries.co.nz</u> www.spaceindustries.co.nz 0800 764 766 (all hours) March 2021

Section 2: Hazards Identification	
HSNO Classification:	5.1.1 B May intensify fire: oxidizer.
Hazard Classification:	6.1D, 8.1A , 8.2C, 8.3A, 9.1A, 9.2 A, 9.3C
	OXIDIZER 5.1

Section 3: Composition/information on ingredients		
Product Description:	Swimming pool chemical, algaecides, biocide, and oxidant.	
	White powder with a slight chlorine odour.	
Components	Calcium Hypochlorite	
CAS Number	7778-54-3	
Proportion	>60%	
Risk Phrases	R20, R21, R22, R31, R34, R41, R50,	

	Section 4: First Aid Measures	
Sr	Show this Safety Data Sheet to a Doctor nort term exposure by all routes is considered to be harmful.	
Inhalation:	Remove victim to fresh air.	
	Seek immediate medical advice.	
Skin Contact:	Immediately flush with large quantities of water.	
	Ensure all contaminated clothing is removed and washed thoroughly.	
Eye Contact:	Irrigate with copious quantities of water for at least 15 minutes.	
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	Eyelids to be held apart.
	Urgently seek medical assistance.
Ingestion:	Immediately remove product from the mouth.
	If swallowed, DO NOT induce vomiting.
	Give a glass of water.
	Contact a Doctor or the Poisons Information Centre (0800 764 766) for further advice
Notes to Doctor:	If swallowed - causes severe burning and corrosion to the mucous membranes and tissues of the mouth, throat and stomach.
	Corrosive to eyes. Can cause corneal burns.
	Skin contact will cause moderate irritation. Corrosive on contact with moist skin and will cause burns.
	If inhaled – mist vapour can produce respiratory irritation and may cause damage of the upper respiratory tract and lung tissues.
For a	dvice, contact the Poisons Information Centre 0800 764 766 or a doctor

	Section 5: Fire Fighting Measures
Specific Hazards:	Non combustible, but will support combustion of other materials
Suitable Extinguishing Media:	Large quantities – Water spray
Fire-fighting advice:	Not combustible, however will support the combustion of other materials. Calcium Hypochlorite is a powerful oxidizing agent and decomposes violently upon heating liberating oxygen, and toxic chlorine gas. In case of fire, area must be evacuated and specialist fire fighters called. Only large quantities of water should be used as an extinguishing agent. If excess water is not available DO NOT attempt to extinguish the fire; use available water to prevent the spread of fire to adjacent property. Attending fire fighters should keep upwind if possible and wear full protective equipment including rubber boots and self-contained breathing apparatus. A fire in the vicinity of Calcium Hypochlorite should be extinguished in the most practical manner, but avoid contaminating this material with the fire fighting agent, including water. Decomposes on contact with water evolving toxic chlorine gas. Once fire is extinguished, wash area thoroughly with excess water. Ensure that drains are not blocked with solid material. Maintenance of excess water during cleaning up operation is essential. Combustible material involved in the incident should be removed to a safe open area for controlled burning or for further drenching with water prior to collection for disposal.

	Section 6: Accidental Release Measures
Procedures to be covered:	<ul> <li>Wear protective equipment to prevent skin and eye contact and breathing in vapours/dust.</li> <li>Air-supplied masks are recommended to avoid inhalation of toxic material.</li> <li>DO NOT return spilled material to original container.</li> <li>DO NOT add small amounts of water to calcium hypochlorite.</li> <li>Sweep up, avoiding generation of dust, then immediately spread as a thin layer in uncontaminated, dry, open area to reduce the possibility of local hot spots forming.</li> <li>Where a spill has occurred in a confined space or an inadequately ventilated enclosure and the material is damp and evolving chlorine, the rate of chlorine evolution can be</li> </ul>



reduced by covering the thinly spread solid with soda ash. For large spills notify the Emergency Services.

	Section 7: Handling and Storage
Handling:	Keep out of reach of children. Read label before use. Wear full protective clothing to avoid splashes.
	No smoking.
Storage:	<ul> <li>Store away from acids, alkalis, reducing agents, detergents or organic materials. Product will react with to produce heat and toxic gases.</li> <li>Keep away from heat.</li> <li>Store in a cool, well ventilated area away from direct sunlight.</li> <li>May be stored in PVC, FRTP, polypropylene or polyethylene containers. Mild steel and stainless steel are rapidly degraded.</li> <li>Copper, brass, bronze and iron will catalytically degrade the product.</li> <li>Storage tanks should be bunded to contain the entire contents in case of leaks or spills.</li> </ul>
	Store away from clothing Keep dry - reacts with water, may lead to drum rupture. Keep containers closed when not in use. Check regularly for spills.

Section 8: Exposure Controls/Personal Protection	
Occupational Exposure Limits:	No value assigned for this specific material by the New Zealand Occupational Safety and Health Service (OSH).
Engineering Control Measures:	Use in a well ventilated area.
	Wear full protective clothing to avoid splashes.
Personal Protective	As product can cause eye irritation, safety glasses or goggles must be worn.
Equipment:	The use of rubber gloves is recommended.
	Wash contaminated clothing and other protective equipment before storage or re-use

Section 9: Physical and Chemical Properties	
Physical state:	Powder
Colour:	White
Odour:	Slight Chlorine
Solubility:	Soluble in water
Specific Gravity/Bulk	
Density:	Approx 0.98 - variable
Vapour Pressure (20 °C):	Not available
Flash Point (°C):	None
Strength:	65% available Chlorine
pH of Solutions:	Alkaline

### Section 10: Stability and Reactivity

Stability:

Powerful oxidizing agent.

Conditions to avoid: Incompatible materials:

Incompatible with Dichloroisocyanuric Acid, Ammonium Nitrate, Trichloroisocyanuric

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Hazardous decomposition products: Hazardous reactions: Acid, or any Chloroisocyanurate.

Calcium Hypochlorite Reacts with water liberating chlorine.

## **Section 11: Toxicological Information**

No adverse health effects expected if the product is handled in accordance with this Safety Data Sheet and the product label. Symptoms or effects that may arise if the product is mishandled and overexposure occurs are:

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mg/kg.
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#### Section 12: Ecological Information

	Avoid contaminating waterways.
Environmental fate, persistence and	
degradation:	This material is biodegradable
Aquatic toxicity:	Very toxic to aquatic organisms. 24hr LC50 (striped bass larvae) = 0.7 mg/L
Terrestrial toxicity:	Expected to be harmful to terrestrial species

### Section 13: Disposal Considerations

Refer to Waste Management Authority. Dispose of material through a licensed waste contractor. Flush to drain with large quantities of water.

Section 14: Transport Information			
Road and Rail Transport:	Classified as a Dangerous Good according to NZS 5433:1999 Transport of Dangerous		
	Goods on Land.		
UN No:	2880		
Class-primary	5.1.1B Oxidizing Agent		
Packing Group:			
Proper Shipping Name:	CALCIUM HYPOCHLORITE, HYDRATED		
Hazchem Code:	2W		
Marine Transport:	Classified as Dangerous Goods by the criteria of the International Maritime Dangerous Goods Code (IMDG Code) for transport by sea; DANGEROUS GOODS		
UN No:	2880		

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Class-primary	5.1.1B Oxidizing Agent
Packing Group:	
Proper Shipping Name:	CALCIUM HYPOCHLORITE, HYDRATED

Section 15: Regulatory Information		
HSNO Classification:	5.1.1 B May intensify fire: oxidizer.	
Hazard Classifications:	6.1 D - Substances which are acutely toxic.	
	8.1 A - Substances that are corrosive to metals.	
	8.2 C - Substances that are corrosive to dermal tissue.	
	8.3 A - Substances that are corrosive to ocular tissue.	
	9.1 A - Substances that are very ecotoxic in the aquatic environment.	
	9.2 A - Substances that are very ecotoxic in the soil environment.	
	9.3 C - Substances that are harmful to terrestrial vertebrates.	

#### Section 16: Other Information

.Issue Date: March 2021

Note: All information given by Space Industries Ltd is offered in good faith and is, to the best of our knowledge, true and accurate. However, since conditions of use are beyond our control, all information relevant to usage is offered without warranty or guarantee and should not be construed as a representation that the product is suitable for any particular purpose or application.